Model Question Paper

Surface Chemistry - Part III

12th Standard

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	Chemistry	Reg.No.:		
ı	.Answer all the questions.		 -	
١	I.Use blue pen only.			
Tin	ne: 01:30:00 Hrs		Total Ma	
	Part-A			5 x 1 = 5
1)	Palladium metal can take up a volume of hydrogen. It is example of			
	(a) dissolution (b) absorbtion (c) adsorbtion (d) desportion			
2)	The multilayer adsorbtion of a gases on solids take place in			
	(a) sols (b) chemisorption (c) physisorption (d) catalyst poison			
3)	A substance which alters the speed of a chemical reaction without itself undergoing any chemical change is called			
	(a) promoter (b) catalyst (c) activator (d) catalytic poison			
4)	The catalyst used for the decompostion of $KCIO_3$ is			
	(a) MnO_2 (b) CI_2 (c) V_2O_5 (d) Pt			
5)	Which one of the following is true about catalyst?			
	(a) Catalyst is used to alter the values of K_p and K_c of an equilibrium reaction (b) A large quantity of catalyst is needed			
	(c) A catalyst alters the position of equilibrium in a reversible reaction (d) A catalyst is specific in its action			
	Part-B		6	x 3 = 18
6)	What is Brownian movement? How is it caused?			
7)	How is the stability of the sol maintained? (or) Give reason for the stab <mark>ility of col</mark> loid.			
8)	How is the stability of the sol maintained? (or) Give reason for the stability of colloid. What is meant by Helmholtz double layer? Define electro-osmosis. Why does sky appear blue? How is mist formed? Part-C Write briefly about the preparation of colloids by dispersion methods. Write notes on i) Dialysis ii) Electrodialysis iii) Ultrafiltration			
9)	Define electro-osmosis.			
10)	Why does sky appear blue?			
11)	How is mist formed?			
	Part-C		5	x 5 = 25
12)	Write briefly about the preparation of coll <mark>oids by d</mark> ispersion methods.			
13)	Write notes on i) Dialysis ii) Electrodialy <mark>sis iii) U</mark> ltrafiltration			
14)	Write notes on i) Brownian movement, ii) Tyndall effect and iii) Helmholtz double layer			
15)	What is electro - osmosis? Explain.			
16)	Write notes on the types of colloids based on the physical states of dispersed phase and dispersion medium.			
	Part-D		2	2X10=20
17)	a) What is electrophoresis? Explain.			
	b) Write notes on emulsions.			
18)	a) How are colloids prepared by using (i) mechanical dispersion method, (ii) electro dispersion method?			
	b) Write notes on a (i) Auto catalyst, (ii) Promoters.			
