Model Question paper Thermodynamics - I 1

11th Standard

	Chemistry	Reg.No.:			T
I. A	nswer all the questions.				-
II. U	Jse blue pen only.				
Time:	00:45:00 Hrs		Tota	l Mark	ເs : 30
	Part - A			4 x	(1=4
1) Wh	hich of the following is not a state function?				
(a)) q (b) q+w (c) Δ H (d) v+pV				
2) Wh	hich of the following is an extensive property?				
(a)	volume (b) density (c) refractive index (d) molar volume				
3) Wh	hich of the following is an exothermic reactions?				
(a)	melting of ice (b) combustion reactions (c) hydrolysis (d) boiling of water				
4) Wh	hich of the following is reversible process?				
(a)	Diffusion (b) melting (c) Neutralization (d) combustion				
	Part - B			3 x	c 2 = 6
5) Na	ame the equipment using which heat of combustion of compounds are determined?				
6) En	ergy can be created and be destroyed.state whether this is true or false.				
7) De	efine zeroth law of thermodynamics.				
	Part - C			5 x 3	3 = 15
8) De	efine an adiabatic process				
9) Wr	rite an difference between an exothermic and an endothermic pr <mark>ocess</mark>				
10) Wh	hat are intensive and extensive properties?				
11) De	fine first law of thermodynamics.				
12) Ex	plain thermal and mechanical equilibrium processes.				
	Part - D			1 ×	c 5 = 5
13) De	escribe a bomb calorimeter and explain <mark>how hea</mark> t of formation of a <mark>n organic</mark> com <mark>pound is d</mark> etermined.				
	- (S) W.				
